LOYOLA COLLEGE (AUTONOMOUS), CHENNAI - 600 034



B.Sc. DEGREE EXAMINATION - **CHEMISTRY**

FIFTH SEMESTER - NOVEMBER 2018

CH 5511- TRANS ELEM. & NUCLEAR CHEMISTRY

Date: 30-10-2018 Dept. No. Max. : 10

Time: 09:00-12:00

PART - A

Answer ALL the questions.

(10x2=20)

- 1. Explain why most of the transition metals form coloured compound.
- 2. What are interstitial compounds? Give an example.
- 3. Differentiate ores and minerals.
- 4. What is electrostatic precipitation?
- 5. What are actinides? Why are they so called?
- 6. Give any two chemical properties of uranium halides.
- 7. Define the term binding energy.
- 8. What are pi-mesons and K-mesons?
- 9. What are radiopharmaceuticals? Give an example.
- 10. Define the term carbon dating.

PART - B

Answer any EIGHT questions.

(8x5=40)

- 11. Discuss the toxic effects of cadmium and mercury.
- 12. Explain the preparation and applications of tungsten bronzes.
- 13. Explain the thermodynamics of reduction processes using Ellingham diagram.
- 14. Explain the mineral beneficiation process.
- 15. How is chromium extracted from its ore?
- 16. What is meant by lanthanide contraction? Explain the consequences of lanthanide contraction on the properties of other group elements.
- 17. Explain the isolation of uranium.
- 18. A freshly cut piece of wood gives 16100 counts of β -ray emission per minute per kg and an old wooden bowl gives 13200 counts per minute per kg. Calculate the age of the wooden bowl. The half-life period of 14 C is 5568 years.
- 19. What is critical mass? How does it explain the principle of atom bomb?
- 20. Define the following with examples. (i) isotopes (ii) isotones (iii) isobars (iv) nuclear isomers.
- 21. Explain neutron activation analysis.
- 22. Discuss the atomic power projects in India.

PART - CAnswer any FOUR questions. (4x10=40)23. (a) Explain the industrial applications of interstitial compounds of titanium and vanadium. **(5)** (b) Explain the magnetic and catalytic property of transition elements. **(5)** 24. (a) Describe how lanthanides are separated by ion-exchange chromatography. **(5)** (b) How is vanadium extracted from its ores? **(5)** 25. (a) What are transactinide elements? How they are significant? **(5)** (b) Explain van Arkel de Boer process. **(5)** 26. (a) Describe the functioning of Scintillation counter. **(5)** (b) Explain n/p ratio. **(5)**

28. (a) The activity of the hair of an Egyptian mammy is 7.0 min⁻¹g⁻¹ of carbon. (5) Find the age of the mummy $(t_{0.5} \text{ of }^{-14}\text{C} = 5770 \text{ years and disintegration rate}$ of freshsample of $^{14}\text{C} = 14 \text{ min}^{-1}\text{g}^{-1}$). (5)

27. (a) Explain natural and induced radioactivity.

(b) Write about the theory of nuclear fission.

(b) Explain the working and applications of a nuclear reactor. (5)

(5)

(5)